Formulation and Evaluation of Sunscreen Gels Containing Mangiferin Isolated from Phaleria macrocarpa Fruits

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ABSTRACT

Objectives: Ultraviolet (UV)-mediated photoreaction and photo-oxidation damage the skin, which can be prevented by using sunscreens as photo protective agents. Mangiferin is a major constituent of Phaleria macrocarpa fruits that has both sunscreen and antioxidant activity. This study aimed to formulate and evaluate sunscreen gel made from mangiferin isolated from Phaleria macrocarpa fruits. Methods: Sunscreen gels were formulated using three different concentrations of mangiferin (1.25%, 2.5%, and 5%) and their physicochemical parameters (color, odor, homogeneity, spread ability, pH, and accelerated stability) were tested. The in vitro Sun Protection Factor (SPF) of the gels was determined using UV spectrophotometry. Sensory evaluation (hedonic test) was performed with a panel of 32 untrained pan- elitists. Skin irritation test was conducted on 20 female volunteers using a skin patch. Results: The three mangiferin sunscreen gels showed high absorbance at wavelengths of 290–360 nm. The SPF was 11.2, 38.6, and and 88.53 at a mangiferin concentration of 1.25%, 2.5%, and 5%, respectively. The gels’ pH was in the proper range (5.8–6.0), and they showed good spread ability, no phase separation, and acceptable consistency. They were found to be stable during a two-month stability study. A gel containing 2.5% concentration of mangiferin is the most preferred formula. In addition, they did not irritate the skin. Conclusion: Gels formulation containing mangiferin at concentrations of 1.25, 2.5, and 5% are effective as sunscreens. The gel meets the requirements on physicochemical parameters and is stable for two months storages at temperatures 8°C, 25°C and 40°C.

Key words: Gel, Sunscreen, Mangiferin, Phaleria macrocarpa, SPF.

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INTRODUCTION

A sunscreen is a photoprotective agent against direct Ultraviolet (UV) radiation and is used as the skin’s defense against the harmful effects of direct UV radiation. Broad-spectrum sunscreens that progressively reduce the harmful effects of direct UV radiation are now being developed.1 Synthetic photoprotective agents can be potentially toxic and carcinogenic and therefore phytoconstituents are becoming popular as essential ingredients of cosmetic formulations because they are natural; have antimutagenic, anticarcinogenic and nontoxic effects; and can severely inhibit the complex process of carcinogenesis. Cosmetics containing natural herbal components are less irritating to the skin, especially skin that is hypoallergic, contains native ingredients and can rejuvenate and adequately protect the skin from environmental pollution, atmospheric temperature fluctuations, UVA and UVB radiation, hyperpigmentation and aging. The use of bioactive compounds in cosmetic formulations has increased in recent years because of their safety, lack of side effects, absence of dangerous synthetic compounds that cause health hazards and ecological sustainability.1,3 Additionally, bioactive compounds have many pharmacological properties; for example, they are natural antioxidants, natural preservatives, hypoallergenic compared to synthetic products and environmentally friendly.4

Phaleria macrocarpa (Mahkota dewa) is a medicinal plant found in Papua, Indonesia. It has long been used as a cure for various diseases such as cancer, liver disease, heart disease, diabetes, arthritis, kidney disease, stroke and hypertension. Besides containing alkaloids, saponins, polyphenols, phenolics glycosides, dodecanoic acid, palmitic acid, ethyl stearate and sucrose,5,6 the Phaleria macrocarpa fruits also contain benzophenone and naturally derived active compounds that possess effective sunscreen activity. Creams and lotions containing the ethanol extract of Phaleria macrocarpa show sunscreen activity in vitro.7 The benzophenone and xanthone glucosides present in the Phaleria macrocarpa fruits are mahkoside A, mangiferin and 6,4-Di-hydroxy-4-methoxybenzophenone-2-O-β-gentiobioside (6,4-DHM P). Natural products, such as mangiferin, of er an innovative solution for modern consumers of cosmetic formulations because they have antiphotoguing, antioxidant and anti-inflammatory properties. Our previous study has shown that mahkoside A, mangiferin and 6,4-DHM P have sunscreen activity in vitro and in vivo.8 Mangiferin at a concentration of 500 µg/mL has a sun protection factor (SPF) of 15.83 and is typically categorized as a sunscreen. Studies have also shown that the application of mangiferin 25% did not cause irritation and allergic reaction with the erythema severity and erythema diameter are both zero.9 Topical application of mangiferin inhibits the increase of skin thickness, wrinkle formation and acute edema on UVB-irradiated mice.9 Mangiferin also shows significant protection against DNA damage and can increase the photoprotective effect of sunscreens.10

Gels are forms of topical dosage that can be properly applied and have excellent stability compared to creams and ointments. Gels also provide controlled release compared to other semisolid formulations.11 Stability testing is a common procedure performed on a cosmetic or drug product. Accelerated stability testing is an initial step that is carried out by choosing various storage temperatures and humidity to evaluate the possibility of product degradation after long-term storage.5 Stability tests of medicinal and cosmetic formulations ensure their strength, quality and purity.
A gel formulation is considered stable if its properties and characteristics stay within acceptable limits for a specific period of normal temperature not exceeding 25°C storage.6

Aim: In this study, we formulated and objectively evaluated sunscreen gels prepared with the use of three different concentrations of mangiferin isolated from P. macaropa fruits. We tested the gels’ physicochemical properties. Additionally, we determined there in vitro SPF using UV spectrophotometry.

Ethics approval: The selection of volunteers and test methods was in accordance with ethical principles specified in the Declaration of Helsinki the study protocol was allowed by the Esa Unggul University Research Ethics committee. For the sensory evaluation and skin irritation test, written informed consent was obtained from all participants before recruitment.

MATERIALS AND METHODS

Materials
Mangiferin isolated from P. macaropa fruits was naturally obtained from the study by Eff et al. (2018).7 Carbopol 934, triethanolamine, propylene glycol, methylparaben, propylparaben, aquadest and methanol were purchased from a local supplier.

Preformulation
Table 1 (Jr Allen, 2009)6 shows the composition of the gel base used in this study. Accurate quantities of carbopol 934, methylparaben and propylparaben were carefully weighed. Carbopol 934 was dispersed into 50 mL of distilled water and stirred to naturally form a gel base. Then, methylparaben and propylparaben already dissolved in propylene glycol were added and the mixture was stirred until it became homogeneous. Mangiferin at three different concentrations of 1.25% (F1), 2.5% (F2) and 5% (F3) was properly introduced into the gel base and the optimum pH was accurately adjusted using triethanolamine.

Evaluation of sunscreen gels
Physicochemical parameters
The physicochemical parameters examined included color, pH, homogeneity, viscosity, spreadability and stability. To accurately measure viscosity, 8 mg each of F1, F2 and F3 gels was centrifuged at 10–100 rpm at 25°C using a Brookfield viscometer (AMETEK Brookfield, Middleboro, MA, USA) with an L4 spindle. To assess spreadability of the gel on the skin, 500 mg each of F1, F2 and F3 gels was placed on transparent glass, which was then covered with another thin glass and the mixture was stirred until it became homogeneous. Mangiferin at three different concentrations of 1.25% (F1), 2.5% (F2) and 5% (F3) was properly introduced into the gel base and the optimum pH was accurately adjusted using triethanolamine.

Table 1: Gel base composition.

<table>
<thead>
<tr>
<th>Ingredients</th>
<th>Function</th>
<th>Percentage (w/w)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Carbopol 934</td>
<td>Gelling agent</td>
<td>1</td>
</tr>
<tr>
<td>Propylparaben</td>
<td>Preservative</td>
<td>0.02</td>
</tr>
<tr>
<td>Methylparaben</td>
<td>Preservative</td>
<td>0.18</td>
</tr>
<tr>
<td>Propylene glycol</td>
<td>Humectant</td>
<td>5%</td>
</tr>
<tr>
<td>Triethanolamine</td>
<td>Surfactant/pH adjuster</td>
<td>until pH was neutral (6-7)</td>
</tr>
<tr>
<td>Aquadest</td>
<td>Solvent</td>
<td>ad 100 ml</td>
</tr>
</tbody>
</table>

Source: Modified from Jr Allen (2009).6

In vitro determination of SPF using UV spectrophotometry
The UV spectrophotometry method used to measure the SPF was adopted from the study by Petro et al. Briefly, 100 mL of methanol was added to 125 mg each of F1, F2 and F3 gels and the mixtures were ultrasonicated for 15 min. Then, 5 mL of each mixture was diluted with methanol up to 50 mL to prepare test solutions. The absorption spectra of the test solutions in a 1 cm cuvette were obtained using an ultraviolet-visible (UV-Vis) spectrophotometer at a wavelength of 290–360 nm. The absorption of the test solutions was measured at intervals of 5 nm and the minimal absorbance equaled 0.05.

log SPF = (AUC/(λn- λ1)) × 2,6 where AUC is the total value of absorbance at λn and λn-1, λ1 is a wavelength of 290 nm.

Sensory evaluation (hedonic test)
A questionnaire was presented to 30 untrained panelists, who provided their opinions on the color, aroma and comfort of F1, F2 and F3 gels.

During assessment, the volunteers were allowed to wash the skin with water only (no soap, detergent, or cosmetic product). T e degree of irritation was assessed using a 4-point scale, depending on the severity of erythema and edema:

- Erythema:
  - 0, no erythema;
  - 1, slight erythema (diameter < 25 mm);
  - 2, erythema clearly visible (diameter 25.1–30 mm);
  - 3, medium erythema (diameter 30.1–35 mm);
  - 4, severe erythema (diameter 35 mm).

- Edema:
  - 0, no edema;
  - 1, slight edema (almost invisible);
  - 2, edema with a clearly defined edge (thickness < 1 mm);
  - 3, mild edema (rising edge ± 1 mm);
  - 4, severe edema (thickness < 1 mm).

During assessment, the volunteers were allowed to wash the skin with water only (no soap, detergent, or cosmetic product). T e irritation index of each of the F1, F2 and F3 gels was calculated using the following formula:
Physicochemical parameter evaluation of gels.

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Color</th>
<th>pH</th>
<th>Homogeneity</th>
<th>Viscosity (cps)</th>
<th>Spreadability (cm)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>50 g</td>
</tr>
<tr>
<td>F1</td>
<td>Pale yellow</td>
<td>6.0</td>
<td>Homogenous</td>
<td>21890</td>
<td>3.5</td>
</tr>
<tr>
<td>F2</td>
<td>Bright yellow</td>
<td>5.8</td>
<td>Homogenous</td>
<td>23270</td>
<td>3.3</td>
</tr>
<tr>
<td>F3</td>
<td>Bright yellow</td>
<td>5.9</td>
<td>Homogenous</td>
<td>29880</td>
<td>3.7</td>
</tr>
</tbody>
</table>

F1: 1.25% mangiferin, F2: 2.5% mangiferin, F3: 5% mangiferin

Physicochemical parameter evaluation results of F1, F2 and F3 gels stored at 8°C, 25°C and 40°C, respectively.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Temperature (°C)</th>
<th>F1</th>
<th>F2</th>
<th>F3</th>
<th>F1</th>
<th>F2</th>
<th>F3</th>
<th>F1</th>
<th>F2</th>
<th>F3</th>
<th>F1</th>
<th>F2</th>
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<tbody>
<tr>
<td>Color</td>
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<td>PY</td>
<td>BY</td>
<td>BY</td>
<td>BY</td>
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<td></td>
<td>40</td>
<td>PY</td>
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<tr>
<td>pH</td>
<td>8</td>
<td>6.0</td>
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<td>Homogeneity</td>
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<td>Phase separation</td>
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</table>

PY, pale yellow; BY, bright yellow; DY, dark yellow; H, homogeneity; -, no change; +, slight change

Table 3: Physicochemical parameter evaluation results of F1, F2 and F3 gels stored at 8°C, 25°C and 40°C, respectively.

Sun Protection factor

Table 4: Sun Protection Factor (SPF) value of gels.

<table>
<thead>
<tr>
<th>Formulation</th>
<th>SPF value</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>11.2</td>
</tr>
<tr>
<td>F2</td>
<td>38.6</td>
</tr>
<tr>
<td>F3</td>
<td>88.33</td>
</tr>
</tbody>
</table>

Sensory evaluation (hedonic test)

Table 5: Sensory evaluation results.

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Color</th>
<th>Consistency</th>
<th>Comfort</th>
<th>Aroma</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>3.65</td>
<td>3.90</td>
<td>3.75</td>
<td>2.95</td>
</tr>
<tr>
<td>F2</td>
<td>2.85</td>
<td>2.95</td>
<td>4.25</td>
<td>3.10</td>
</tr>
<tr>
<td>F3</td>
<td>3.60</td>
<td>3.60</td>
<td>4.05</td>
<td>3.25</td>
</tr>
</tbody>
</table>

Skin irritation

Safety testing is essential before raw materials or end products can be sold to consumers. The degree of irritation for all 20 volunteers in this study was 0, indicating that gels F1, F2 and F3 do not irritate the skin.

DISCUSSION

Development of sunscreen preparations leading to the use of natural materials today is preferred because the public more simply accepts it. People assume natural ingredients are safer to use and less negative impact than chemicals. Potential natural material as sunscreen is Phalleria macrocarpa. P. macrocarpa contains the main compound, which is a benzophenone derivative that possesses a protective effect against the dangers posed by ultraviolet light.

Table 2 shows gel product and the physicochemical parameter evaluation results of F1, F2 and F3 gels, respectively. The gel formula were found to be homogeneous. Freshly prepared F1 gel was pale yellow, while F2 and F3 gels were bright yellow. The pH of the F1, F2 and F3 gels ranged from 3.6 to 6.0, which indicates they are not irritating.
from 5.8 to 6.0, corresponding to the human skin pH of 4.5–7. So, if applied to the skin, the three gels would not cause irritation. Viscosity is a measure of a fluid’s resistance to flow: the higher the viscosity, the greater the resistance to flow. Viscosity is an important parameter for evaluating gel preparations. In this study, F1, F2 and F3 gels had a viscosity of 2,000–50,000 cps, meeting the requirements of sunscreen preparations. Spread ability indicates the extent to which a gel spreads when applied to skin and the therapeutic potential of a gel formulation depends on this spread ability. In this study, the gels F1, F2 and F3 had good spread ability in the range of following 3.5–4.4 for F1, 3.3–4.8 for F2 and 3.7–5.2 for F3.

Table 3 shows the physicochemical parameter evaluation results of F1, F2 and F3 gels stored at various temperature for 8 weeks. Physical stability tests determined the physical changes that occurred in gels F1, F2 and F3. Physical stability tests are associated with shelf life in storage of a gel formulation and are essential to ensure the quality, safety and efficacy of the product. T. et al. tests play an important role in the development and improvement of formulations, determining validity and monitoring physical and chemical characteristics. T. et al. test parameters evaluated for physical stability were color, pH, homogeneity and phase separation. Organoleptic observations at storage temperatures of 8°C, 25°C and 40°C for 8 weeks did not show any changes in color, pH, homogeneity and phase separation in F1 and F2 gels. However, we observed a few changes in color and phase separation in gel F3 at weeks 7 and 8; the color changed from pale yellow to dark yellow, and there was slight phase separation presumably due to the oily phase separation promoted at a higher temperature.

The effectiveness of F1, F2 and F3 gels was tested by determining their SPF in vitro using the method developed in the study by M. et al. (1986). F1, F2 and F3 gels showed high absorbance at 290–360 nm wavelength. Table 4 represents Sun Protection Factor (SPF) value of gels. T. et al. SPF was 11.2, 38.6 and 88.53, respectively, indicating that all three gels have sunscreen activity. Increasing the mangiferin might increase the SPF because of increased amounts of phenolic compounds, which have a protective effect due to the presence of double bonds that are conjugated with a single bond and are involved in the absorption of sunlight.

P. marocarpa fruits have been traditionally used as medicines, either singly or mixed with other traditional medicines. New compounds discovered in these fruits are benzophenones and xanthine glycosides: mahoside A; mangiferin; and 6,4-DHMP. In vitro and in vivo tests of these compounds show that they act as sunscreens. M. angiferin is a polyphenol compound contained in leaves (Mangifera indica) and fruits (P. marocarpa) and has antioxidant properties. At a concentration of 100 ppm, mangiferin has a SPF of 2.82 and, at concentrations of 12.5%, 25% and 50%, it decreases erythema severity and erythema diameter significantly even compared to negative controls (p < 0.05). Some phytoconstituents, such as proanthocyanidin, quercetin, apigenin, silymarin and carotenoids are potential ingredients of sunscreens. Phenolic compounds found in specific plants are potential agents for typically preventing the harmful effects of direct UV radiation on skin. Evidence suggests that various forms of polyphenols used both orally and topically are beneficial for skin health and prevent sunburn. Polyphenols can be used as a possible alternative for skin care and effective protection against the harmful effects of direct UV radiation, but large-scale clinical studies are still required in order to adequately assess the use of polyphenols in effective prevention of sunburns, both topically and orally. Active polyphenolic compounds, such as flavonoids, have striking similarities with organic UV filters that possess chromophores and aromatic rings having not only photo protective effects but also anti-oxidant activity. Free radicals produced endogenously during cellular metabolism or exogenously sourced from UV radiation and pollution can damage the skin at levels both cellular and tissue. Although the body contains endogenous antioxidants to prevent free-radical damage, this system has limited capacity and leads to oxidative stress, triggering carcinogenesis. Using topical preparations with antioxidant activity can neutralize exogenous and endogenous oxygen species.

Sensory evaluation / hedonic test results (Table 5) represent F1 gel was the most preferred formulation on the parameters of color and consistency, while F2 and F3 gels were most preferred on the parameters of comfort and aroma, respectively. T. et al. test is the most widely used test to measure the level of preference for a product. Results from skin irritation test showed that mangiferin sunscreen gel works as a surfactant, so it prevents skin irritation. M. angiferin has anti-inflammatory and antioxidant activities. It inhibits the expressions of Interleukin 6 (IL-6) and IL-1β, decreases total inflammatory cell cell infiltration and eosinophils and lowers prostaglandin (PG) D2. M. angiferin also inhibits Immunoglobulin E (IgE) production, anaphylaxis, histamine-induced vascular permeability, histamine release and the lymphocyte proliferative response.

CONCLUSION

Gel formulations containing mangiferin at concentrations of 1.25%, 2.5% and 5% are potential and active as sunscreens with Sun Protection Factor (SPF) values of 11.2, 38.6 and 88.53, respectively. All three gel formulas meet the physicochemical parameter and stable at temperatures 8°C, 25°C and 40°C for eight weeks of storage. A gel containing 2.5% concentration of mangiferin is the most preferred formula. T. et al. gel formulas were tolerated because they did not irritate respondents.

ACKNOWLEDGEMENT

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CONFLICT OF INTEREST

T. et al. authors declare no conflict of interest.

ABBREVIATIONS

SPF: Sun Protection Factor; DHMP: dihydroxy-4-methoxybenzophenone-2-O-β-D-glucopyranoside.

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